

**2,2'-Bipyridyl-2-ium aqua(2,2'-bipyridyl- $\kappa^2N,N'$ )-tetakis(nitrato- $\kappa^2O,O'$ )praseodymium(III)-2,2'-bipyridyl hemisolvate**Iraida A. Charushnikova<sup>a\*</sup> and  
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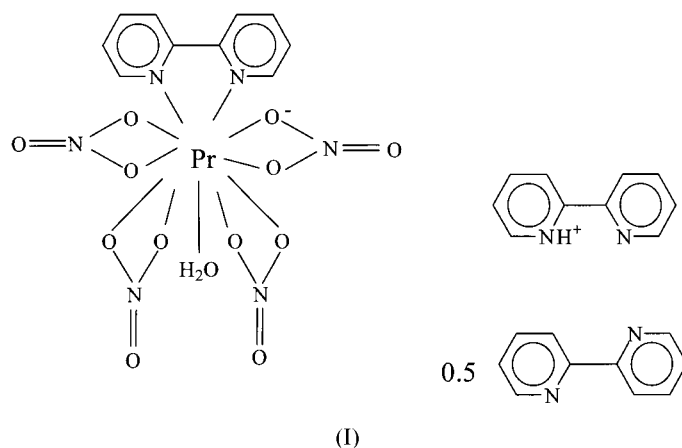
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**Key indicators**Single-crystal X-ray study  
 $T = 130$  K  
Mean  $\sigma(C-C) = 0.005$  Å  
 $R$  factor = 0.029  
 $wR$  factor = 0.067  
Data-to-parameter ratio = 12.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The Pr atom in the title complex,  $(C_{10}H_9N_2)[Pr(NO_3)_4 \cdot (C_{10}H_8N_2)(H_2O)] \cdot 0.5C_{10}H_8N_2$ , is 11-coordinated. Hydrogen bonds between the coordinated water molecule, the uncoordinated 2,2'-bipyridylium  $[(HBipy)^+]$  and the centrosymmetric 2,2'-bipyridyl (Bipy) solvent molecules link the structural components into layers parallel to the  $ab$  plane.

**Comment**

The title compound,  $(HBipy)[Pr(NO_3)_4(Bipy)(H_2O)] \cdot 0.5Bipy$  (Bipy is 2,2'-bipyridyl), (I), is a new complex characterized by a high coordination number of Pr. The Pr atom is eleven-coordinated (Fig. 1). The 2,2'-bipyridyl solvent molecules lie on crystallographic inversion centers.

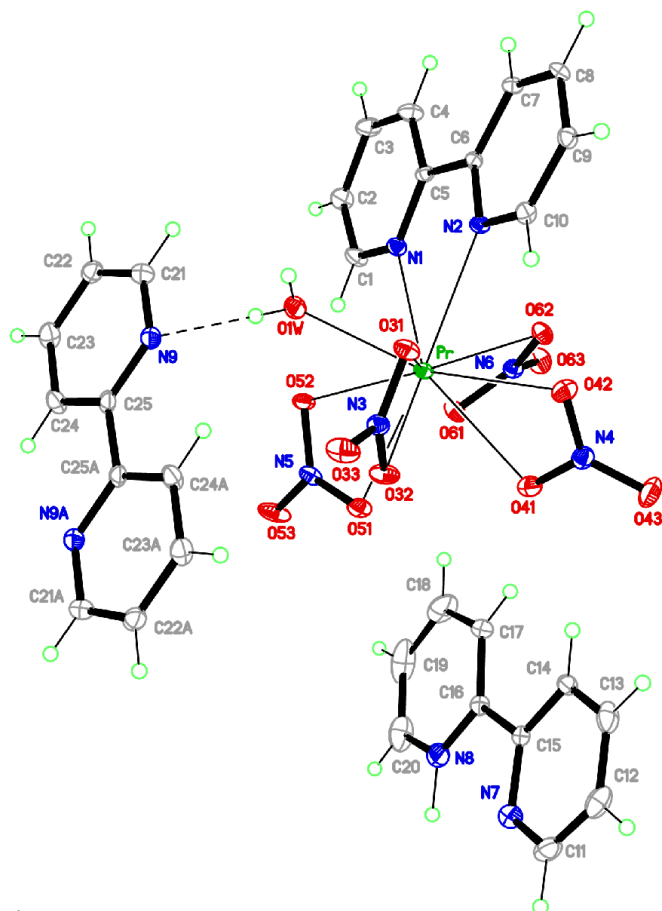


Among the Bipy complexes of lanthanides, the coordination number of 11 has been found only for  $[La(NO_3)_3(Bipy)(H_3COH)(H_2O)_2] \cdot C_{10}H_{20}O_5$ , (II) (Zhen & Rogers, 1994). Four nitrate anions in the  $[Pr(NO_3)_4(Bipy)(H_2O)]^-$  complex anion of (I) are coordinated in bidentate mode. One of the Pr–O(nitrate) bonds is significantly elongated [2.752 (2) Å] in comparison to the seven others [average 2.612 (2) Å]. A similar elongation of one of the La–O(nitrate) bonds was found also in the neutral  $[La(NO_3)_3(Bipy)(H_3COH)(H_2O)_2]$  complex of (II) (Zhen & Rogers, 1994). Several types of hydrogen bonds,  $O_w-H \cdots O(\text{nitrate})$ ,  $O_w-H \cdots N(\text{Bipy})$  and  $N(\text{Bipy})-H \cdots O(\text{nitrate})$  link all the structural components of (I). Each  $(HBipy)^+$  cation acts as proton donor to form a hydrogen bond with a coordinated nitrate anion. The coordinated water molecule forms two hydrogen bonds. The first is with a terminal atom of one nitrate anion of a neighboring complex; two such bonds form a centrosymmetric dimeric complex (Fig. 2). The second is with a solvent Bipy molecule. Each solvent Bipy molecule has two hydrogen

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**Figure 1**  
ORTEP/ORTEP-3 (Burnett & Johnson, 1996; Farrugia, 1997) view of (I), showing 30% probability displacement ellipsoids. The dashed line indicates a hydrogen bond.

bonds of this type, linking the dimers into infinite chains. The chains form layers parallel to the *ab* plane (Fig. 2).

## Experimental

The crystals were obtained by slow evaporation at 293 K of a xylene solution (1 ml) containing  $\text{Pr}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$  (0.05 mol) and Bipy (the  $\text{Pr}^{3+}$ –Bipy molar ratio is 1:1).

### Crystal data

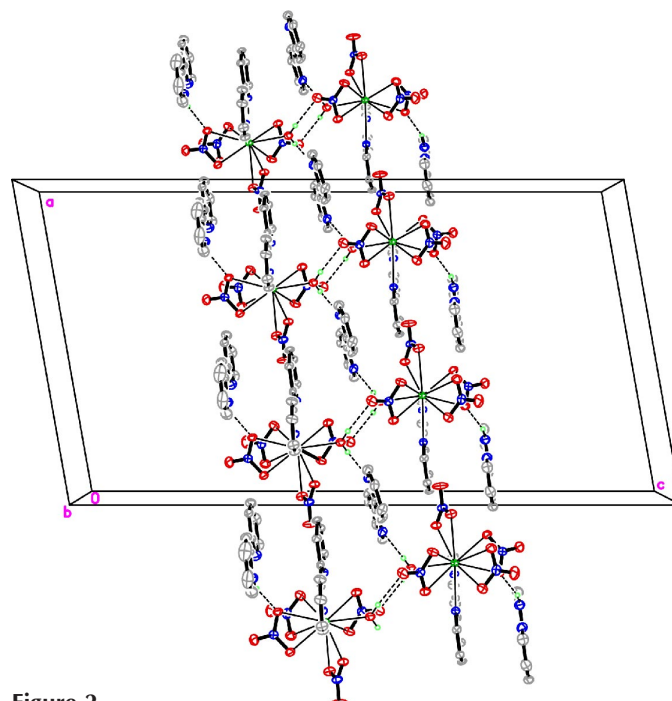
$(\text{C}_{10}\text{H}_8\text{N}_2)_2[\text{Pr}(\text{NO}_3)_4(\text{C}_{10}\text{H}_8\text{N}_2) \cdot (\text{H}_2\text{O})] \cdot 0.5\text{C}_{10}\text{H}_8\text{N}_2$   
 $M_r = 798.43$   
 Monoclinic,  $C2/c$   
 $a = 17.7935$  (4) Å  
 $b = 10.2712$  (3) Å  
 $c = 32.8924$  (8) Å  
 $\beta = 100.0016$  (11)°  
 $V = 5920.1$  (3) Å<sup>3</sup>  
 $Z = 8$

### Data collection

Nonius KappaCCD diffractometer  
 $\varphi$ – $\omega$  scans  
 Absorption correction: multi-scan (MULABS; Blessing, 1995)  
 $T_{\min} = 0.638$ ,  $T_{\max} = 0.732$   
 15166 measured reflections  
 5632 independent reflections

$D_x = 1.792$  Mg m<sup>−3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 15 166 reflections  
 $\theta = 2$ –27.5°  
 $\mu = 1.73$  mm<sup>−1</sup>  
 $T = 130$  (2) K  
 Prism, green  
 0.36 × 0.22 × 0.18 mm

5000 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.045$   
 $\theta_{\max} = 27.5^\circ$   
 $h = -21 \rightarrow 21$   
 $k = -13 \rightarrow 12$   
 $l = -33 \rightarrow 42$



**Figure 2**  
CAMERON (Watkin *et al.*, 1993) view of the packing of (I), showing the hydrogen-bonding interactions as dashed lines.

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.029$   
 $wR(F^2) = 0.067$   
 $S = 1.13$   
 5632 reflections  
 442 parameters

H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0144P)^2 + 13.7668P]$$

where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.52 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.64 \text{ e \AA}^{-3}$

**Table 1**

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$\text{N8}-\text{H8} \cdots \text{O62}^i$	1.07 (4)	2.03 (4)	3.028 (4)	155 (3)
$\text{O1W}-\text{H12} \cdots \text{N9}$	0.75 (4)	2.00 (4)	2.751 (3)	177 (4)
$\text{O1W}-\text{H11} \cdots \text{O33}^{ii}$	0.73 (4)	2.09 (4)	2.809 (3)	171 (4)

Symmetry codes: (i)  $\frac{1}{2} + x, y - \frac{1}{2}, z$ ; (ii)  $\frac{1}{2} - x, \frac{1}{2} - y, -z$ .

All H atoms on carbon were positioned geometrically ( $C-H = 0.93$  Å) and refined using the riding model [ $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ ]. H atoms of the water molecule were located in difference Fourier maps and refined with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ . The H atom on nitrogen was refined with a fixed displacement parameter [ $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$ ].

Data collection: *KappaCCD Software* (Nonius, 1997); cell refinement: *HKL* (Otwinowski & Minor, 1997); data reduction: *HKL*; program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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